



Analysis of Limestone using the Spectrolaser

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Limestone, the ubiquitous natural form of calcium carbonate, is used extensively worldwide as a feedstock in the glass, ceramic, and cement industries. To ensure production quality in these industries, the level of impurities in limestone and other like feedstock is closely monitored during the production process.

The Spectrolaser is a convenient tool to analyze feedstock materials such as limestone where broad elemental range and fast analysis aids the production process.

Sample Preparation

Typically, the Spectrolaser is used to analyze pressed powders which give a measurement precision of <5% RSD¹ for most matrices. This precision level is limited by the heterogeneous properties of naturally occurring materials and is similar to other direct analysis techniques such as XRF. Precision can further be enhanced to <1% RSD by the preparation of the samples in fused beads prior to analysis.

¹ RSD: Relative Standard Deviation is defined as the standard deviation in a series of measurements divided by the mean value of the measurement.

Detectable Elements

The Spectrolaser records the laser plasma emission spectrum from the deep UV (190 nm) to the IR (950 nm) using a patented multi-channel spectrometer system that captures the entire spectral range in a single laser pulse. This enables simultaneous determination of all detectable elements. A broad range of elements are detectable in limestone depending on the geology of the particular source of the material. Detectable elements include, but are not confined to, Al, Ba, C, Ca, Cr, Fe, H, K, Mg, Mn, Na, O, Si, Sr, and Ti.

Calibration Curves

The Spectrolaser uses a technique known as Laser Induced Breakdown Spectroscopy (LIBS) to perform the elemental analysis of materials. Calibration involves the use of certified reference materials, the selection of an appropriate elemental optical emission line, and, in most cases, selection of a normalization emission line. The Spectrolaser software automatically constructs calibration curves of the normalized peak area vs. elemental concentration present in the reference materials. The concentration

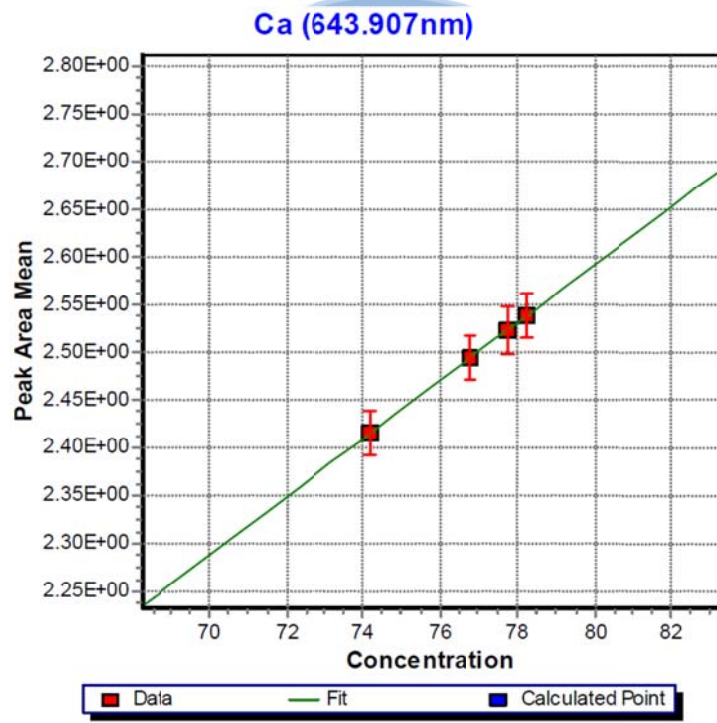
is determined in samples of unknown composition with the comparison against this calibration reference.

When there is a dominant matrix component for which the concentration will remain approximately constant across the calibration set, precision can be improved by normalizing against an emission line corresponding to the optical emission from that element. In the case of limestone, calcium or carbon are possible candidates for matrix normalization. Alternatively, since the analysis is performed in air, atmospheric components, for example Nitrogen and Oxygen, will also be present in constant proportion and can be used in this role.

Example Calibration Curves

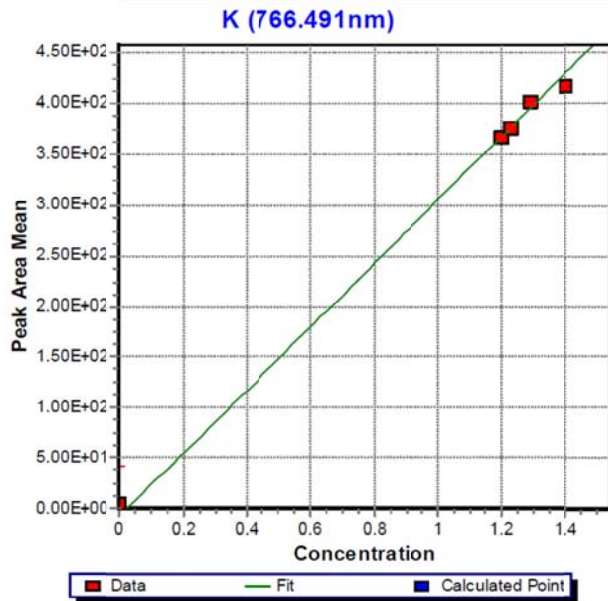
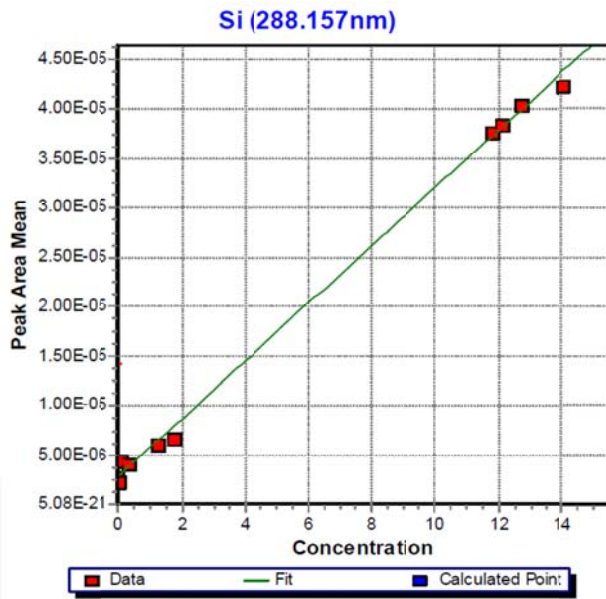
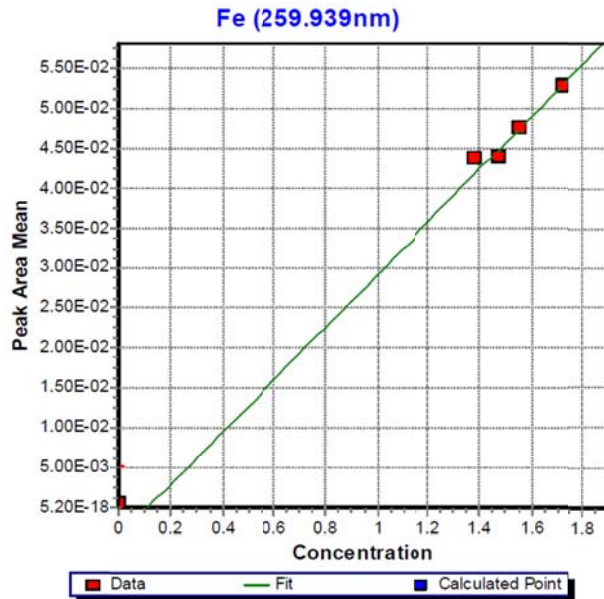
Principal Component

CaO 70 – 80%



Example Calibration Curves of Minor Components

Fe, K, and Si



Reproducibility

A typical limestone analysis using pressed pellets:

	MgO	Al ₂ O ₃	SiO ₂	SO ₃	K ₂ O	Fe ₂ O ₃
Measurement	293nm	309.27nm	263.1nm	PLS	766.491nm	259.939nm
1	1.4	4.6	12.3	0.22	1.27	1.47
2	1.3	4.7	11.8	0.26	1.19	1.47
3	1.6	4.5	12.0	0.26	1.20	1.45
4	1.5	5.0	12.2	0.26	1.19	1.46
5	1.4	4.2	11.9	0.21	1.24	1.42
6	1.5	4.7	12.0	0.22	1.22	1.47
7	1.6	4.5	12.2	0.22	1.25	1.44
8	1.4	4.0	11.9	0.21	1.26	1.41
Average (SD)	1.5 (0.1)	4.5 (0.3)	12.0 (0.2)	0.23 (0.02)	1.23 (0.03)	1.45 (0.02)
Reference	1.55	4.62	12.14	0.21	1.23	1.47

Typical Detection Limits – Calcium Matrices

Element	Detection Limit*
Na	0.001 %
Ca	0.001 %
Mg	0.001 %
Fe	0.003 %
Al	0.003 %
K	0.002
Ti	0.002
Si	0.003%